

Study on electrochemical properties of titanium-oxide $\text{Ru}_{0.3}\text{La}_x\text{Ti}_{(0.7-x)}$ electrode*

Chunlian Luo¹²³, Jianhua Wu¹², Yanling Yang¹²

1. College of Marine Engineering, Jimei University, Xiamen 361000, China, E-mail:
clianluo@163.com(C.L.);

2. Key Laboratory for Marine Corrosion and Intelligent Protection Materials of Xiamen,
Jimei University, Xiamen 361000, China

3. Navigation College, Xiamen Ocean Vocation College, Xiamen 361000, China

Abstract The $\text{Ru}_{0.3}\text{La}_x\text{Ti}_{(0.7-x)}$ electrodes are made by sol-gel method. The effect on chlorine evolution performance, strengthening life and surface morphology of electrodes doped with La_2O_3 are studied. Results: The titanium etched by acid and then alkali has a large specific surface area, while the titanium etched by acid had no obvious nanopore; The incorporation of La_2O_3 could reduce the film resistance and electrochemical transfer resistance of the electrode, enhance the conductivity of the coating; The electrode doped with 1% La_2O_3 coating has a smaller electrochemical transfer resistance than Commercial DSA under the same unit molar mass of Ru, which will reduce the cost of Ru; The electrode doped with 1% La_2O_3 has the highest reaction rate constant among the electrode doped with 0% La_2O_3 , 2% La_2O_3 and 3% La_2O_3 ; The electrode doped with 1% La_2O_3 has higher life than the electrode doped with 2% La_2O_3 and 3% La_2O_3 .

Keywords: AC impedance; Metal oxide electrode; Chlorine anode; Mechanism; Accelerated life