

Photo-induced passivation: A new corrosion mitigation strategy for bronze artifacts

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Abstract To explore the potential protective effect of UV illumination on the tin bronzes, the corrosion behaviors of four ternary Cu-Sn-Pb alloys with varying tin contents in a chloride-containing borate buffer solution (pH 8.4) were investigated in both the dark and UV illuminated conditions. UV illumination was found to consistently inhibit bronze corrosion across varying tin contents by facilitating the formation of a compact SnO₂ layer on the surfaces. This process is referred to as photo-induced passivation. Concerning the effect of Sn content on the photo-induced corrosion process, a substantial Sn enrichment within the patina layer favors the formation of this compact oxide film more readily. Moreover, the oxide films on high-tin bronzes generally exhibit greater photostability and provide more long-lasting protection compared with low-tin bronzes. Two kinds of artificial patinated bronzes were prepared to further access the inhibitive effect of photo-induced passivation treatment on the bronzes covered with typical patina layers. The maximum inhibition efficiencies achieved were 93.1% for the patinated Cu₅Sn bronze, which is covered with a Sn-rich single-layered patina mainly consisting of cuprite, and 82.4% for the patinated Cu₁₀Sn bronze, which features a Cu₂O/CuCl inner layer and a Cu₂(OH)₃Cl outer layer. These values are comparable to those achieved by organic substances used on various artificially patinated bronzes. Overall, photo-induced passivation treatment can effectively mitigate the corrosion of Cu-Sn-Pb alloys without introducing foreign substances that might interfere with the archaeological information of bronze artifacts during the preservation process. However, further research is still necessary before it can be applied to actual bronze artifacts.

Keywords bronze; photo-induced passivation; patina; oxide film

Reference

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